

	AHSAA Homeschool Student Eligibility Exams	
		Percentage
Standard	Standard Text	of Test
Reference		Items
Matter and Its Interactions		50%
	Use the periodic table as a model to predict the relative properties and	
	trends (e.g., reactivity of metals; types of bonds formed, including ionic,	
	covalent, and polar covalent; numbers of bonds formed; reactions with	
1	oxygen) of main group elements based on the patterns of valence	
1	electrons in atoms.	
3	reaction involving main group elements.	
	Analyze and interpret data using acid-base indicators (e.g., color-changing	
	markers, pH paper) to distinguish between acids and bases, including	
4	comparisons between strong and weak acids and bases.	
	Use mathematical representations to support and verify the claim that	
_	atoms, and therefore mass, are conserved during a simple chemical	
5	reaction.	
6	Develop models to illustrate the concept of half-life for radioactive decay.	
6	Research and communicate information about types of naturally occurring	
6.a	radiation and their properties.	
Ch	Develop arguments for and against nuclear power generation compared to	
0.0	Chekility Fores and Interactions	40%
Initial and stability: Forces and interactions		4070
	hasic concents of distance displacement speed velocity and acceleration	
	(e.g., velocity versus time graphs, displacement versus time graphs,	
7	acceleration versus time graphs).	
	Apply Newton's laws to predict the motion of a system by constructing	
	force diagrams that identify the external forces acting on the system,	
	including friction (e.g., a book on a table, an object being pushed across a	
8	floor, an accelerating car).	
	Use mathematical equations (e.g., (m1v1 + m2v2) before = (m1v1+ m2v2)	
	after) and diagrams to explain that the total momentum of a system of	
9	objects is conserved when there is no net external force on the system.	
	Use the laws of conservation of mechanical energy and momentum to	
9.8	predict the result of one-dimensional elastic collisions.	
	batteries and apply Obm's law to solve typical problems demonstrating the	
10	effect of changing values of resistors and voltages.	



Standard Reference	Standard Text	Percentage of Test Items
Energy; Waves and Their Applications in Technologies for Information Transfer		10%
11	Design and conduct investigations to verify the law of conservation of energy including transformations of potential energy, kinetic energy, thermal energy, and the effect of any work performed on or by the system.	
12	Design, build, and test the ability of a device (e.g., Rube Goldberg devices, wind turbines, solar cells, solar ovens) to convert one form of energy into another form of energy.	
13	Use mathematical representations to demonstrate the relationships among wavelength, frequency, and speed of waves (e.g., the relation v = lambda f) traveling in various media (e.g., electromagnetic radiation traveling in a vacuum and glass, sound waves traveling through air and water, seismic waves traveling through Earth).	
14	Propose and defend a hypothesis based on information gathered from published materials (e.g., trade books, magazines, Internet resources, videos) for and against various claims for the safety of electromagnetic radiation.	
15	Obtain and communicate information from published materials to explain how transmitting and receiving devices (e.g., cellular telephones, medical- imaging technology, solar cells, wireless Internet, scanners, Sound Navigation and Ranging [SONAR]) use the principles of wave behavior and wave interactions with matter to transmit and capture information and energy.	